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SECTION 3.1

MEASUREMENTS

AND THEIR

UNCERTAINTY (pages

63–72) This section

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describes the concepts of accuracy, precision, and error in measurements. It also explains the proper use of significant figures in measurements and calculations.

~~SECTION 3.1
MEASUREMENTS
AND THEIR
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Measurements and
Their Uncertainty

OBJECTIVES:

–Distinguish among
accuracy, precision,
and error of a
measurement. 4

Section 3.1

Measurements and
Their Uncertainty

OBJECTIVES:

–Determine the
number of significant
figures in a

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measurements and in
a calculated answer.

5 Measurements

Qualitative

measurements are
words, such as heavy
or hot

~~Chapter 3~~

~~Measurements and~~

~~Their Scientific~~

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When a

measurement

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Reported as 5.0 kg is divided by 3.0 L, for example, the display may show

1.666666667 as the answer. We are justified in reporting the answer to only two significant figures, giving 1.7 kg/L as the answer, with the last digit understood to have some uncertainty.

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significant figures in
the following
measurements, 3.85 x
b. 17.30 cmS perform
the operations and
give in Standard form
With the number of
significant 37.2mL.
8.382 Ck. b. a.1Z9Y10
= 2.1

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accuracy, Precision,
and error Accuracy,

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Measurements, and error help determine the reliability of measurements.

The accuracy of a measurement is determined by how close the measured value is to the actual value. The precision of a measurement is determined by how close repeated

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Measurements are to
one another.

Error is the
difference between
the measured value
and the accepted
value.

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The uncertainty can
also be stated as a

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percentage of the measured value. This is called the percentage uncertainty, $\%$. This can be calculated by taking the absolute uncertainty and dividing it by the mean, or measured value as below.
$$\% \text{ uncertainty} = \frac{\text{absolute uncertainty}}{\text{measured value}} \times 100$$

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