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ENERGY—HEAT AND

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505–510) This section

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explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.

Energy Transformations
(page 505)

*SECTION 17.1 THE
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(pages 505–510)*

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(pages 505-510) This

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process that takes in

heat. exothermic. a

chemical or physical

process that releases

heat. enthalpy. the heat

content of a system. heat

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energy required to
change a substance from
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of 5500.2 kJ when 1.0
mol is combusted at 1.0
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internal energy of
5512.4 kJ ...

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