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EQUATIONS:

*Students using
models.*

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Academy 03—

~~Algebraic Method~~

~~for Balancing~~

~~Chemical~~

~~Equations~~

~~(Example #2)~~

~~Molecular Models~~

~~of the~~

~~Functional~~

~~Groups and Fatty~~

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**Beginner's Guide
to Balancing
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the Algebraic

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(algebraic

method) |

Homework Tutor

~~Balancing~~

~~Equations Using~~

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~~Models Quick~~

Use models as a means for balancing

chemical equations.

Materials . For each group of 4 students: • Cup of Skittles • 2 Paper Plates .

Procedure . 1. Label one plate reactants and

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Balancing

Equations

the other products, to represent each side of a

chemical

equation. Use at least four different colors of skittles (representing atoms of different elements) to make models of

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the substances
in each equation
below.

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~~Balancing~~

~~Chemical~~

~~Equations Using~~

~~Models~~

1. Use
toothpicks and
gumdrops of at
least four
different colors
(representing

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atoms of

(different
elements) to

make models of
the substances

in each equation
below. 2. For

each reaction
below, use your

models to
determine the

products needed
(if not already

indicated), and

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then balance the
equation.

Using Models

Quick Lab

~~1B, 8C Balancing
Equations by~~

~~Using Models~~

The equation is:

$H_2 + O_2$ yields

H_2O My daughter

thought about

this and

envisioned using

pom-pom craft

balls of various

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sizes to represent the atoms of hydrogen and oxygen in the equation. After a quick trip to Wal-Mart for supplies, the following activity is what resulted from her efforts.

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~~Balancing~~

~~Chemical~~

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~~a Tactual Model~~

~~Answers~~

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Equations by

Using

Mathematical

Model Step 1,

Write down your

given equation.

For this

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example, you

will use: $C_3H_8 + O_2 \rightarrow H_2O + CO_2$

This reaction

occurs when

propane (C_3H_8)

is burned in the

presence of

oxygen to

produce water

and carbon

dioxide. Step 2,

Write down the

number of atoms

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per element .

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Sodium metal
reacts with
water to form
sodium hydroxide
solution and

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hydrogen gas.

Write a balanced equation for the reaction,

including state symbols. Reveal answer

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~~chemical~~

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Equation

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balancing
chemical
equations for
they are so
easy!” said no
one ever. Sure,
you may feel
like it when

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In this lesson,

students are

introduced to

the idea of

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balancing
chemical

reaction

equations and

conservation of

matter by using

molecular

models. Students

are developing

and using models

(SEP 2) in order

to fully

understand

conservation of

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matter—in order for a chemical reaction to make products, the atoms needed must have come from somewhere. They are building the reactant and product molecules, and then determining if they have

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balanced the
numbers of each
type of atom.

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This answer can
be checked by
substituting $(y = -2)$ into the
original

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equation. $[4y + 5 = 4 \times -2 + 5 = -8 + 5 = -3]$ The

equation

balances, so $(y = -2)$ is the final answer.

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